

# New Ring-Expansion Reactions of Hydroxy Propenoyl Cyclic Compounds under Palladium(0)/Phosphine-Catalyzed Conditions

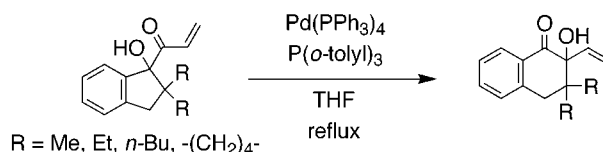
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## ABSTRACT



Palladium(0)-catalyzed one-atom ring-expansion of 1-hydroxy-2,2-dialkyl-1-propenoylindane derivatives has been achieved in the presence of  $P(o\text{-tolyl})_3$  giving 2-hydroxy-3,3-dialkyl-2-vinyl-1-tetralone derivatives in excellent yields. This ring-expansion reaction was applied to a 17-(1-oxo-2-propenyl)- $\beta$ -estradiol derivative and furnished a similar ring-expanded product in an excellent yield.

Ring-expansion reactions have provided chemists a considerably useful tool for the construction of various biologically active natural products and drugs.<sup>1</sup> Recent efforts in our laboratory have focused on base-mediated and Pd(0)-catalyzed ring-expansion reactions ( $5 \rightarrow 7$  and  $5 \rightarrow 6$ ) using various hydroxy allenyl cyclic compounds.<sup>2</sup> Liebeskind and Stone<sup>3a</sup> and the Butenschön group<sup>3b</sup> reported acid ( $CF_3CO_2H$ )-catalyzed ring-expansion reactions ( $4 \rightarrow 5$ ) of methoxyallenylcyclobutenols via corresponding enone interme-

diates, thus releasing four-membered ring strain.<sup>3a</sup> However, a similar acid-catalyzed ring-expansion reaction of the 1-hydroxy-2,2-dimethyl-1-propenoylindane **2a** with  $CF_3CO_2H$  resulted in a very low yield (6%) of a ring-expanded product **3a** and 89% recovery of **2a**. Conventional reaction conditions using  $BF_3 \cdot OEt_2$  and KOH for the well-known  $\alpha$ -ketol rearrangement<sup>4</sup> were also not available for the ring-expansion reactions of **2a**. We describe here the first successful demonstration of the Pd(0)- and phosphine-catalyzed one-atom ring-expansion reactions of **2a–d** and **9** under neutral conditions.

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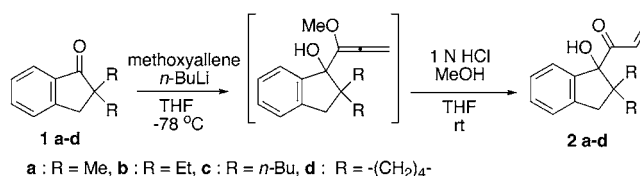
<sup>‡</sup> Rigaku Corporation.

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## Scheme 1



**Table 1.** One-Atom Ring-Expansion Reactions of 1-Hydroxy-2,2-dialkyl-1-propenoylindans **2a–d**

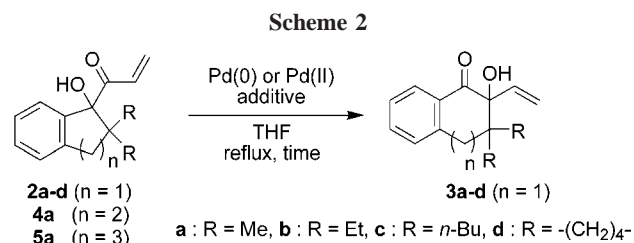
entry	<b>2</b>	Pd (mol %)	additive (mol %)	time (h)	product	yield (%) <sup>a</sup>
1	<b>2a</b>	Pd(PPh <sub>3</sub> ) <sub>4</sub> (5.0)		7	<b>3a</b>	52 <sup>b</sup> (40 <sup>b</sup> ) <sup>c</sup>
2	<b>2b</b>	Pd(PPh <sub>3</sub> ) <sub>4</sub> (5.0)		15	<b>3b</b>	68 <sup>b</sup> (22 <sup>b</sup> ) <sup>c</sup>
3	<b>2a</b>	Pd(PPh <sub>3</sub> ) <sub>4</sub> (5.0)	P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	7	<b>3a</b>	92
4	<b>2b</b>	Pd(PPh <sub>3</sub> ) <sub>4</sub> (5.0)	P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	15	<b>3b</b>	99
5	<b>2c</b>	Pd(PPh <sub>3</sub> ) <sub>4</sub> (5.0)	P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	18	<b>3c</b>	90
6	<b>2d</b>	Pd(PPh <sub>3</sub> ) <sub>4</sub> (5.0)	P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	22	<b>3d</b>	80
7	<b>2a</b>		P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	7	<b>3a</b>	43 <sup>b</sup> (29 <sup>b</sup> ) <sup>c</sup>
8	<b>2a</b>		P( <i>o</i> -tolyl) <sub>3</sub> (30.0)	24	<b>3a</b>	71
9	<b>2a</b>		PPh <sub>3</sub> (10.0)	7	<b>3a</b>	55
10	<b>2a</b>		P( <i>n</i> -Bu) <sub>3</sub> (10.0)	7	<b>3a</b>	12
11	<b>2a</b>		DPPE (10.0)	7	<b>3a</b>	13
12	<b>2a</b>		DABCO (10.0)	7	<b>3a</b>	23 <sup>b</sup> (trace <sup>b</sup> ) <sup>c</sup>
13	<b>2a</b>		DBU (10.0)	7	<b>3a</b>	53
14	<b>2a</b>		Et <sub>3</sub> N (10.0)	7	<b>3a</b>	25 <sup>b</sup> (45 <sup>b</sup> ) <sup>c</sup>
15	<b>2a</b>		Me <sub>2</sub> S (10.0)	7	<b>3a</b>	13 <sup>b</sup> (71 <sup>b</sup> ) <sup>c</sup>
16	<b>2b</b>		P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	15	<b>3b</b>	62 <sup>b</sup> (27 <sup>b</sup> ) <sup>c</sup>
17	<b>2a</b>	PdCl <sub>2</sub> (MeCN) <sub>2</sub> (5.0)		24	<b>3a</b>	26 <sup>b</sup> (44 <sup>b</sup> ) <sup>c</sup>
18	<b>2a</b>	PdCl <sub>2</sub> (MeCN) <sub>2</sub> (5.0)	P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	24	<b>3a</b>	28 <sup>b</sup> (13 <sup>b</sup> ) <sup>c</sup>
19	<b>2a</b>	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub> (2.5)		24	<b>3a</b>	5 <sup>b</sup> (67 <sup>b</sup> ) <sup>c</sup>
20	<b>2a</b>	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub> (2.5)	P( <i>o</i> -tolyl) <sub>3</sub> (10.0)	7	<b>3a</b>	83

<sup>a</sup> Isolation yield. <sup>b</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) analysis. <sup>c</sup> Recovery of starting material.

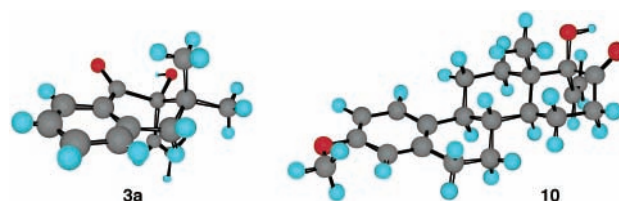
The precursor compounds **2a–d** were prepared in 75–88% yields by treatment of the corresponding dialkylindanones **1a–d**<sup>5</sup> with lithio methoxyallene, followed by hydrolysis with 1 N HCl in MeOH–THF (Scheme 1).<sup>6,7</sup>

First, **2a** was refluxed for 7 h in the presence of 5 mol % Pd(PPh<sub>3</sub>)<sub>4</sub> alone in THF. The desired one-atom ring-expansion reaction proceeded to afford the 2-hydroxy-3,3-dialkyl-2-vinyl-1-tetralone derivative **3a** in 52% yield, but **2a** (40%) was recovered (entry 1 in Table 1). Similar treatment of **2b** for 15 h gave **3b** in 68% yield, together with 22% recovery of **2b** (entry 2 in Table 1). These results suggested the decomposition of the Pd catalyst under the reaction conditions used. Thus, **2a** was refluxed for 7 h in the presence of 5 mol % Pd(PPh<sub>3</sub>)<sub>4</sub> and 10 mol % P(*o*-tolyl)<sub>3</sub> as an additive,<sup>8</sup> which is known to stabilize Pd catalysts and improve turnover in THF. The desired ring-expansion reaction proceeded smoothly to afford **3a** in 92% yield (entry 3 in Table 1). Then, similar reactions of **2b–d** using 5 mol % Pd(PPh<sub>3</sub>)<sub>4</sub> and 10 mol % P(*o*-tolyl)<sub>3</sub> furnished **3b** in 99% yield, **3c** in 90% yield, and **3d** in 80% yield, respectively (entries 4–6 in Table 1). All experimental results are summarized in Scheme 2 and Table 1.

The structures of all ring-expanded products **3a–d** were explicitly determined by their characteristic spectroscopic data and/or by X-ray crystallographic analysis of **3a**, as shown in Figure 1.<sup>7,9</sup> Treatment of the six- and seven-



membered compounds **4a** and **5a**, respectively, with 5 mol % Pd(PPh<sub>3</sub>)<sub>4</sub> and 10 mol % P(*o*-tolyl)<sub>3</sub> in THF under reflux



**Figure 1.** Computer-generated drawing from the X-ray coordinates of compounds **3a** and **10**.

for 24 h led only to recovery of each corresponding starting compound in either 79 or 90% yield.

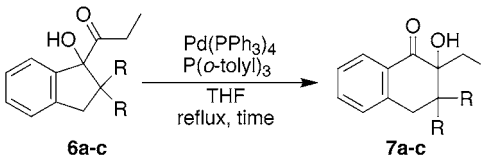
To clarify the ring-expansion mechanism, several reactions were carried out as follows. Compounds **2a,b** were allowed

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 (7) See also Supporting Information.  
 (8) For review, see: Crisp, G. T. *Chem. Soc. Rev.* **1998**, 27, 427.

(9) X-ray data for **3a**: C<sub>14</sub>H<sub>16</sub>O<sub>2</sub>, MW = 216.28, colorless needle, monoclinic, space group *P*2<sub>1</sub>/*c* (#14), *a* = 13.2235(4) Å, *b* = 10.1349(5) Å, *c* = 18.0755(5) Å, *V* = 2298.0(2) Å<sup>3</sup>, β = 108.443(4)°, *Z* = 8, *R* = 0.131, *R*<sub>w</sub> = 0.061. Structure factors are available from author upon request.

to react with several Lewis bases [e.g., P(*o*-tolyl)<sub>3</sub>, PPh<sub>3</sub>, 1,2-bis(diphenylphosphino)ethane (DPPE), DABCO, DBU, Et<sub>3</sub>N, Me<sub>2</sub>S], which are commonly employed in the traditional Morita–Baylis–Hillman reaction,<sup>10</sup> in THF under reflux without the use of Pd(PPh<sub>3</sub>)<sub>4</sub> to give **3a** or **3b** in 12–71% yields, respectively (entries 7–16 in Table 1). On the basis of recent reports<sup>11</sup> that the propenoyl moiety can coordinate to Pd(II), **2a** was treated with PdCl<sub>2</sub>(MeCN)<sub>2</sub> alone or a mixture of PdCl<sub>2</sub>(MeCN)<sub>2</sub> and P(*o*-tolyl)<sub>3</sub>. These reactions gave **3a** in 26 and 28% yields, respectively, together with the recovery of **2a** (entries 17 and 18 in Table 1). Upon treatment with Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> alone as a Pd(0) catalyst, the reaction of **2a** scarcely proceeded (entry 19 in Table 1), but in the presence of P(*o*-tolyl)<sub>3</sub>, the yield of **3a** was improved to 83% (entry 20 in Table 1). Thus, it was demonstrated that the Pd(0) species were essential catalysts, and the Pd(0)-catalyzed ring-expansion reaction required an efficient additive such as P(*o*-tolyl)<sub>3</sub>. Subsequently, we investigated the importance of the C=C bond in the propenoyl moiety of **2a–c**. 1-Hydroxy-1-propanoylindan derivatives **6a–c**, obtained by the catalytic hydrogenation of **2a–c**, were subjected to the same reaction conditions as described above, but the corresponding α-tetralone derivatives **7a–c** were obtained in scant amounts, as shown in Table 2. When **6a** was refluxed in the presence of 50 mol

**Table 2.** Attempt at Ring-Expansion Reactions of 1-Hydroxy-1-propanoylindans **6a–c**



a: R = Me  
b: R = Et  
c: R = *n*-Bu

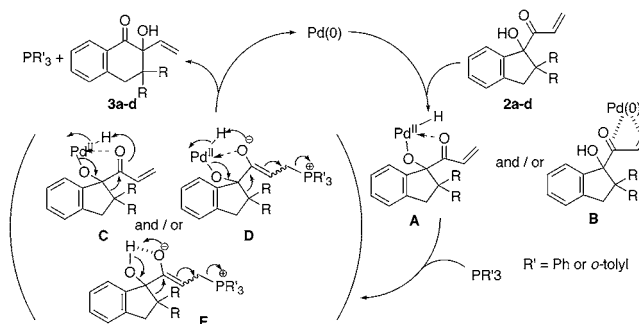
entry	<b>6</b>	Pd(PPh <sub>3</sub> ) <sub>4</sub> (mol %)	P( <i>o</i> -tolyl) <sub>3</sub> (mol %)	time (h)	product	yield (%) <sup>a</sup>
1	<b>6a</b>	5	10	7	<b>7a</b>	10 (87) <sup>b</sup>
2	<b>6a</b>	50	100	24	<b>7a</b>	88 <sup>c</sup>
3	<b>6b</b>	5	10	15	<b>7b</b>	12 (78) <sup>b</sup>
4	<b>6c</b>	5	10	18	<b>7c</b>	8 (92) <sup>b</sup>

<sup>a</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) analysis. <sup>b</sup> Recovery of starting material. <sup>c</sup> Isolation yield.

% Pd(PPh<sub>3</sub>)<sub>4</sub> and 100 mol % P(*o*-tolyl)<sub>3</sub> in THF, **7a** could be obtained in 88% yield. These results suggested that the presence of the C=C bond in the conjugated propenoyl moiety of **2a–c** was very efficient in the Pd(0)- and P(*o*-tolyl)<sub>3</sub>-catalyzed ring-expansion reactions.

Speculative mechanisms for the Pd(0)- and phosphine-catalyzed ring-expansion reactions can be suggested on the

basis of the experimental results described above (Figure 2).<sup>2f,10–13</sup> In the first step of the catalytic cycle, the oxidative



**Figure 2.** Plausible mechanisms for the Pd(0)- and phosphine-catalyzed ring-expansion reactions of **2a–d**.

addition of the hydroxy group of **2a–d** to a Pd(0) catalyst generates the complex **A**,<sup>2f,11–13</sup> in which the reductive elimination of the Pd(II) species, followed by migration of the C1–C2 bond in a concerted manner (**C**), may give the ring-expanded products **3a–d**.

Conjugate addition of the phosphine species to the propenoyl moiety of **A** may form complex **D**,<sup>10</sup> in which synchronous release of the Pd(II) species and the phosphonium moiety, followed by the migration of the C1–C2 bond, would afford **3a–d**. In another plausible first step of the catalytic cycle, the coordination of the propenoyl moiety of **2a–d** to the Pd(0) species may generate the weak complex **B**, whereby the easy conjugate addition of the phosphine species<sup>10</sup> may generate the phosphonium intermediate **E**. In **E**, hydrogen abstraction from the C1–OH group by the resulting enolate, followed by ketonization and release of the phosphonium moiety, would occur, together with synchronous double-bond migration and C1–C2 bond migration to give **3a–d**.

The β-ketol rearrangement reaction of 17-hydroxy-20-ketosteroids, referred to as the *D*-homo rearrangement, was discovered in 1938 and has been the subject of investigation ever since its discovery.<sup>14</sup> Although all rearrangement reactions of steroidal compounds are typically performed by employing various bases and Lewis acids,<sup>14</sup> such a neutral Pd(0)- and phosphine-catalyzed rearrangement reaction has never been reported. Hence, we attempted the ring-expansion reaction of 17-(1-oxo-2-propenyl)-β-estradiol derivative **9**

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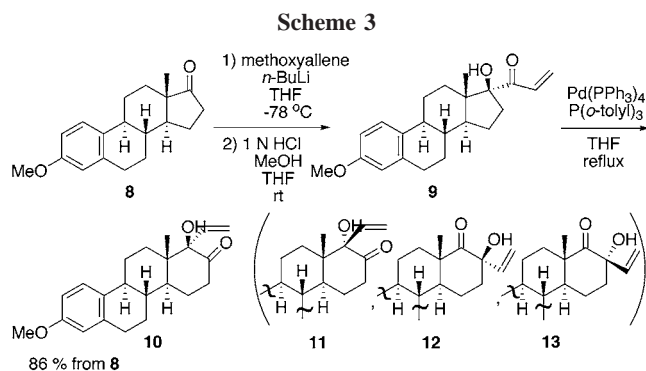
(11) (a) Hosokawa, T.; Shinohara, T.; Ooka, Y.; Murahashi, S. I. *Chem. Lett.* **1989**, 2001. (b) Kawatsura, M.; Hartwig, J. F. *Organometallics* **2001**, *20*, 1960. (c) Miller, K. J.; Kitagawa, T. T.; Abu-Omar, M. M. *Organometallics* **2001**, *20*, 4403.

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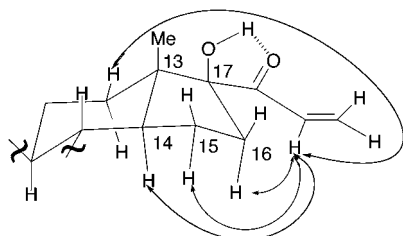
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(vide infra), which was prepared from **8**, as shown in Scheme 3. Crude compound **9**, without being purified on a silica gel



column, was refluxed in the presence of 5 mol %  $\text{Pd}(\text{PPh}_3)_4$  and 10 mol %  $\text{P}(o\text{-tolyl})_3$  in THF for 24 h. The desired ring-expansion reaction proceeded in a stereospecific manner to afford the sole product **10** in 86% yield from **8** among four possible products, **10**–**13** (Scheme 3). The structure of **10** was clarified by X-ray crystallographic analysis, as shown in Figure 1.<sup>7,15</sup>

On the basis of the stereochemistry ( $^1\text{H}$  NOESY experiment shown in Figure 3) of the precursor **9** and the product



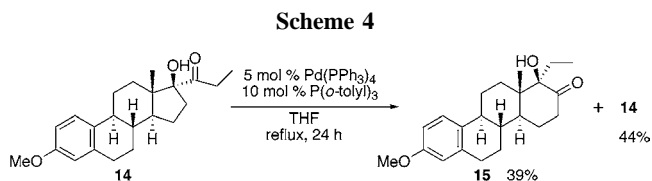
**Figure 3.** Selected  $^1\text{H}$  NMR NOE enhancement for **9**.

**10**, it becomes evident that the ring-expansion reaction proceeded by the stereospecific migration of the C13–C17 bond rather than by that of the C16–C17 bond.

(15) X-ray data for **10**:  $\text{C}_{22}\text{H}_{28}\text{O}_3$ , MW = 340.46, colorless plate, orthorhombic, space group  $P2_12_12_1$  (#19),  $a = 7.4067(3)$  Å,  $b = 7.8333(2)$  Å,  $c = 30.5900(9)$  Å,  $V = 1774.80(10)$  Å<sup>3</sup>,  $Z = 4$ ,  $R = 0.065$ ,  $R_w = 0.115$ . Structure factors are available from author upon request.

To confirm the importance of the conjugated propenoyl system in the  $\text{Pd}(0)$ - and phosphine-catalyzed ring-expansion of **9**, compound **14**, which was obtained by hydrogenation of **9**, was refluxed for 24 h under the same conditions as those used in the case of **9**. The ring-expansion reaction proceeded to afford **15** in 39% yield, but 44% of **14** was recovered (Scheme 4). This result suggested that the C=C bond of the conjugated propenoyl moiety played an important role; thus, the Morita–Baylis–Hillman reaction<sup>10</sup> appeared to be involved in the  $\text{Pd}(0)$ - and phosphine-catalyzed ring-expansion reaction. The structure and stereochemistry of **15** were determined by its identification with the product, which was prepared by catalytic hydrogenation of **10**.<sup>7</sup> Thus, the  $\text{Pd}(0)$ - and phosphine-catalyzed ring-expansion reaction of **9** can be similarly explained in terms of the case of **2a–d**, as shown in Figure 2.

In conclusion, we demonstrated new  $\text{Pd}(0)$ - and  $\text{P}(o\text{-tolyl})_3$ -catalyzed one-atom ring-expansion reactions of 1-hydroxy-2,2-dialkyl-1-propenoylindan derivatives and a steroidal compound. These new reactions are expected to be useful for the construction of cyclohexanones bearing vinyl and hydroxy groups and for further allylic rearrangement.<sup>16</sup>



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**Supporting Information Available:** Typical experimental procedure for the synthesis of compounds **1b,c**, **2a–d**, **3a**, **4a**, **5a**, **6a–c**, **7a**, **9**, **10**, **14**, and **15** and their physical and spectroscopic data, as well as X-ray structural data of compounds **3a** and **10** (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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